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strong acid is titrated
with a weak base, the
pH of the solution at
the equivalence point
is than 7. 19. A is a
highly purified solid
used to check the
concentration of a
standard solution. 20.

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pH of the solution at
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is than 7 19 A is a
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States of Matter PART I
In the space provided,
write the letter of the
term or phrase that
best completes each
statement or best
answers each question.

_____ 1. According to
the kinetic-molecular
theory, particles of an
ideal gas a. attract
each other but do not

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CHAPTER 10 REVIEW

States of Matter

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the

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crystal c covalent
molecular crystal (b) is
hard, brittle, and
nonconducting a
metallic crystal (c)
typically has the lowest
melting point of the
four

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Ms. Agostine's
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Particle Mass number

19. Proton 20. Neutron

21. Electron 10

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continued SHORT
ANSWER Write the
answers to the
following questions in
the space provided. 22.

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CHAPTER 9 REVIEW
Stoichiometry MIXED
REVIEW SHORT
ANSWER Answer the
following questions in

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the space provided. 1.
Given the following
equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO_2(g) + 2H_2O(g)$
a. What is the value of the coefficient x in this equation?
40.07 g/mol
b. What is the molar mass of C_3H_4 ?
2 mol O_2 :1 mol H_2O
c. What is the mole ratio ...

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CHAPTER 1 REVIEW
Matter and Change
Page 24/26

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MIXED REVIEW SHORT

ANSWER Answer the following questions in the space provided. 1.

Classify each of the following as a homogeneous or heterogeneous substance.

homogeneous a. sugar

homogeneous d.

plastic wrap

homogeneous b. iron

filings heterogeneous

e. cement sidewalk

heterogeneous c.

granola bar 2. For each

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appropriate ...

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